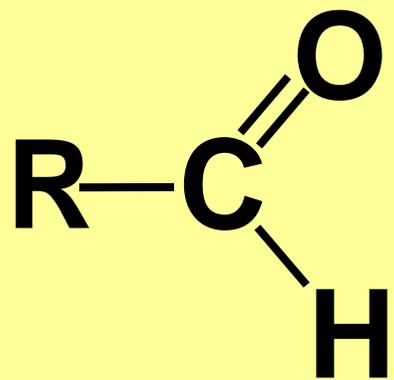
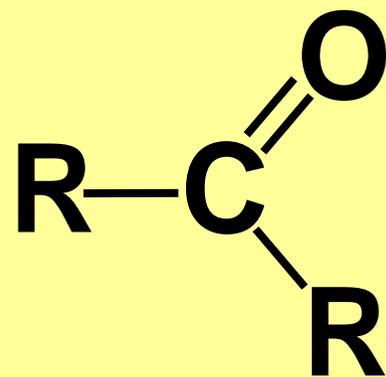


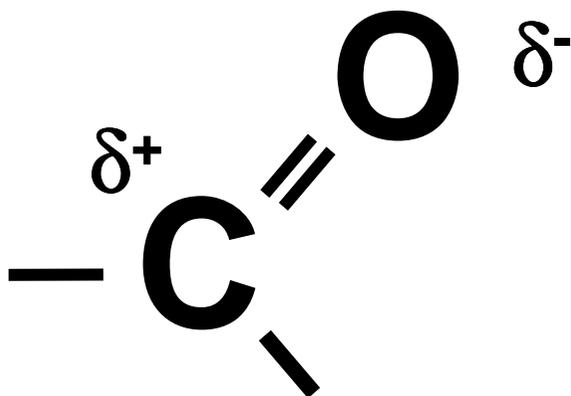
Aldeidi e Chetoni



aldeide

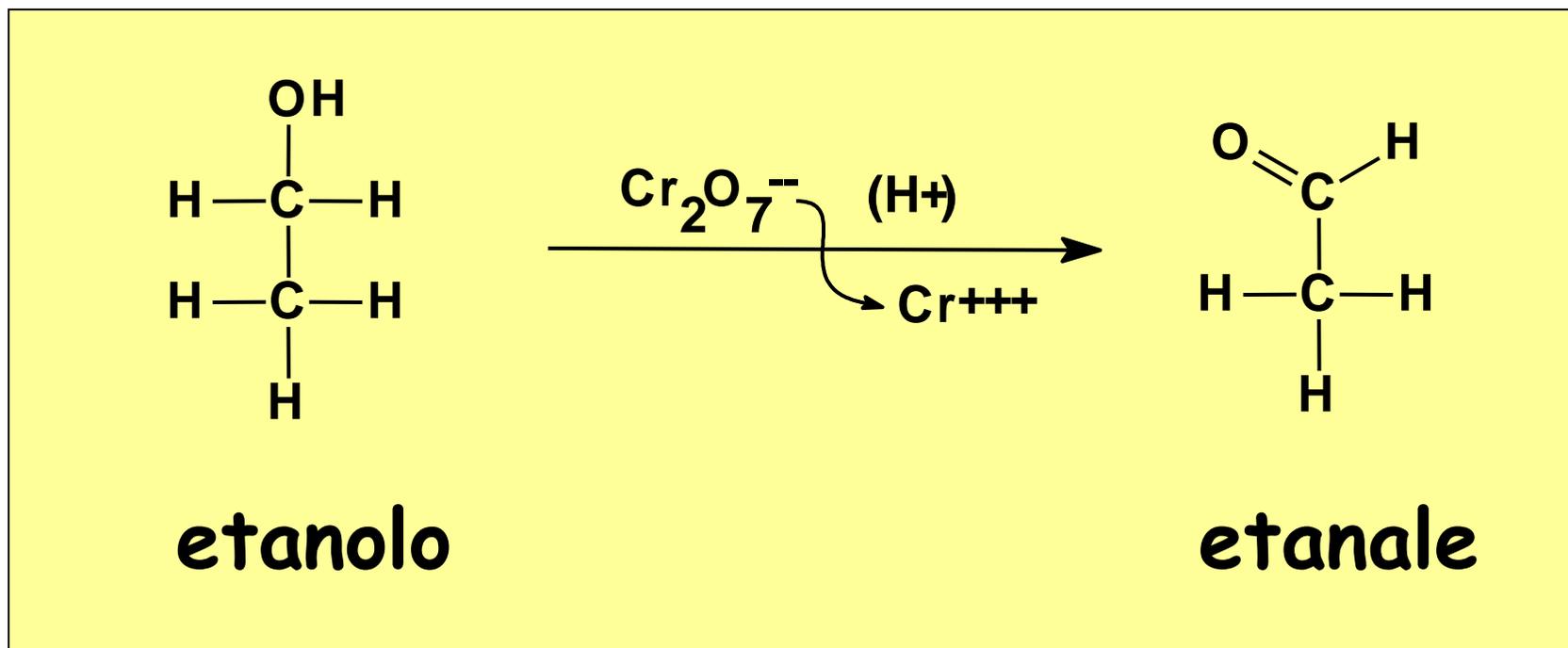


chetone

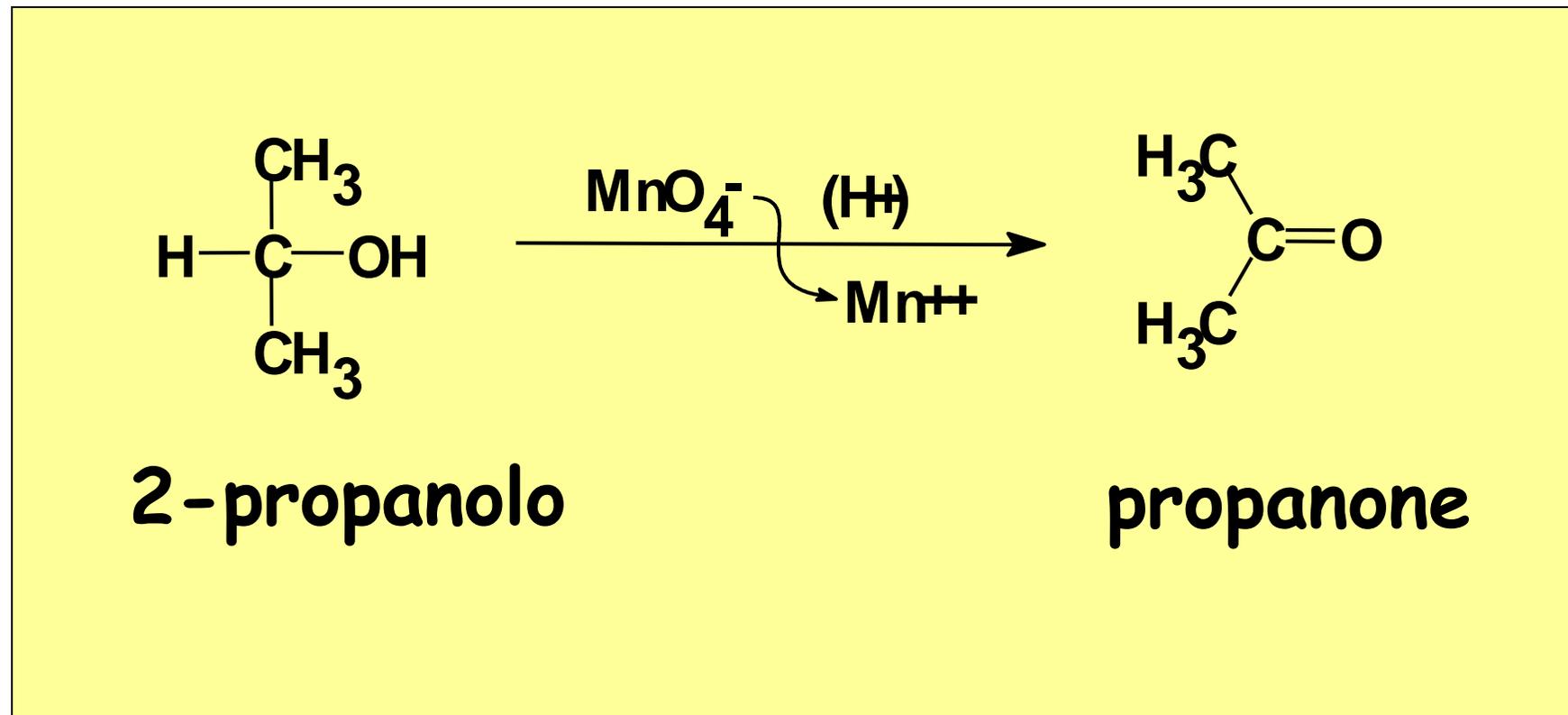


Il gruppo, comune alle due classi, è detto gruppo carbonilico o carbonile; tale gruppo è fortemente polarizzato e in esso il C è ibridato sp^2

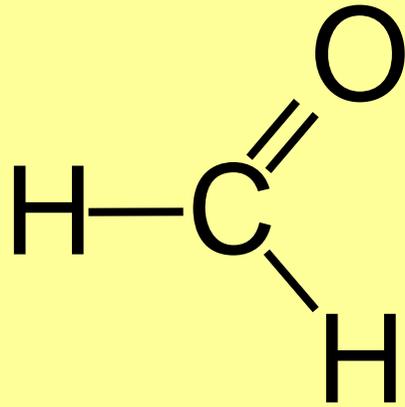
Le aldeidi si ottengono per ossidazione degli alcoli primari.



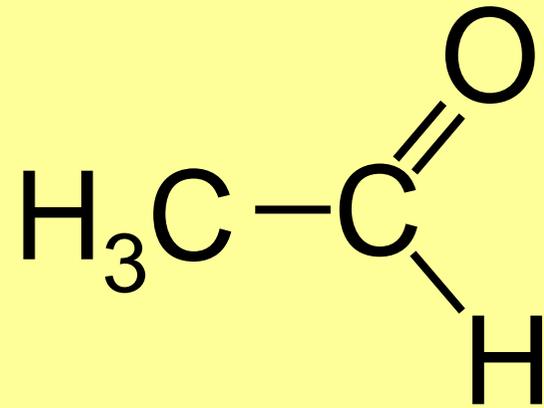
I chetoni si ottengono per ossidazione degli alcoli secondari.



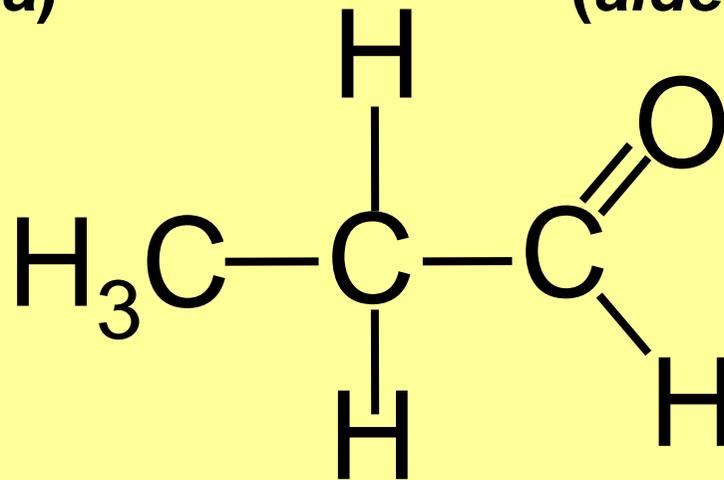
Nomenclatura



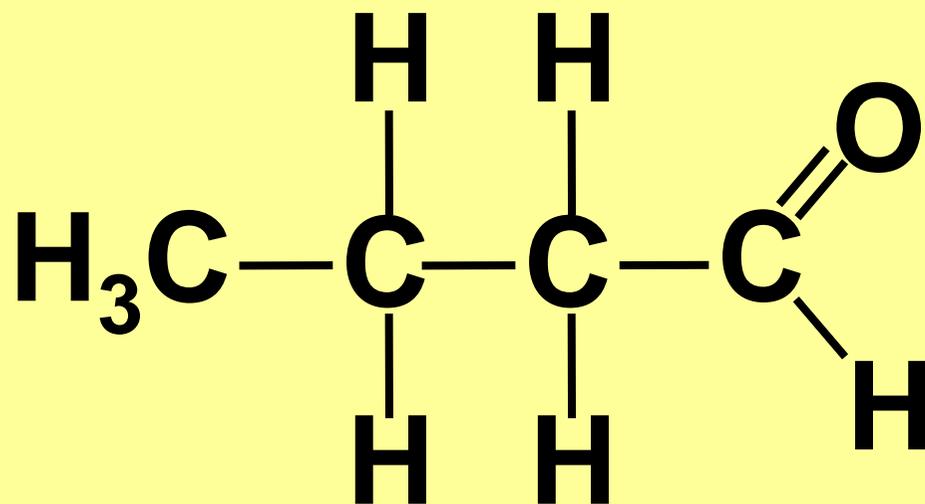
metanale
(aldeide formica)



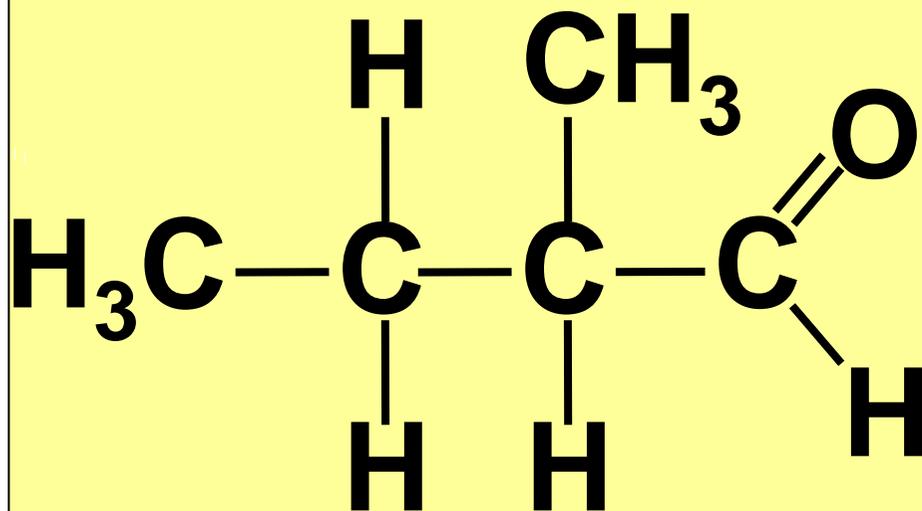
etanale
(aldeide acetica)



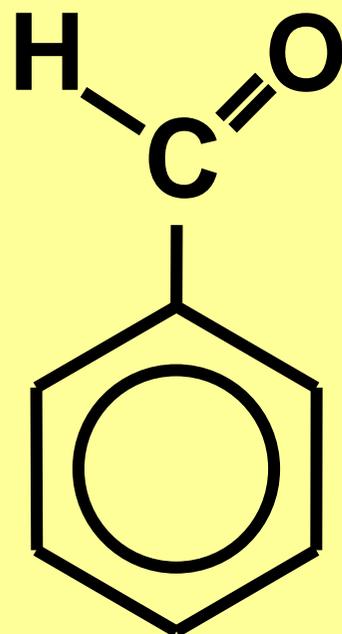
propanale
(aldeide propionica)



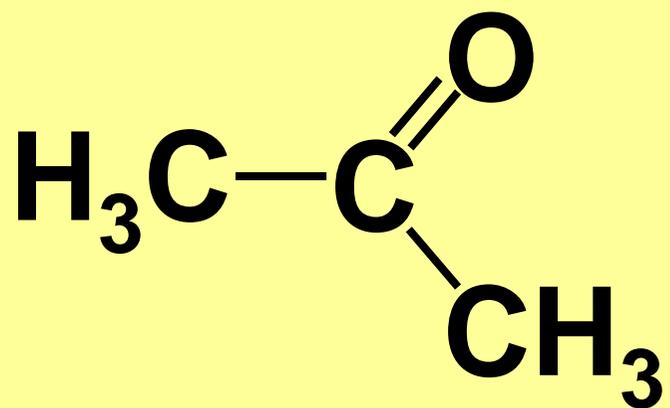
butanale
(aldeide butirrica)



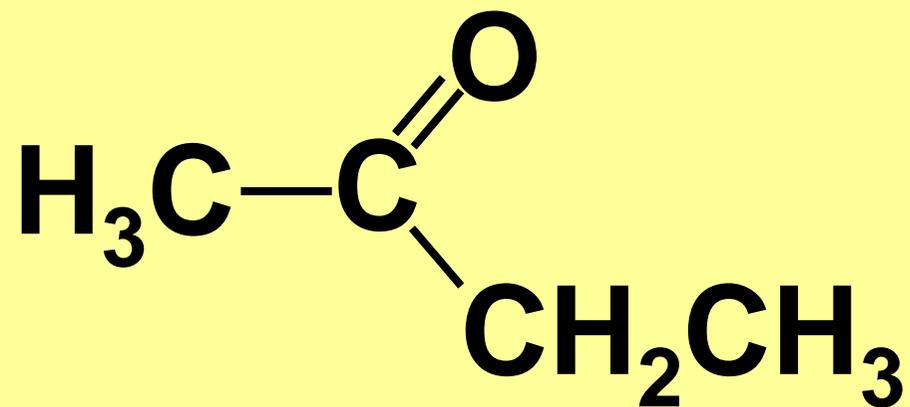
2-metilbutanale
(aldeide 2-metilbutirrica)



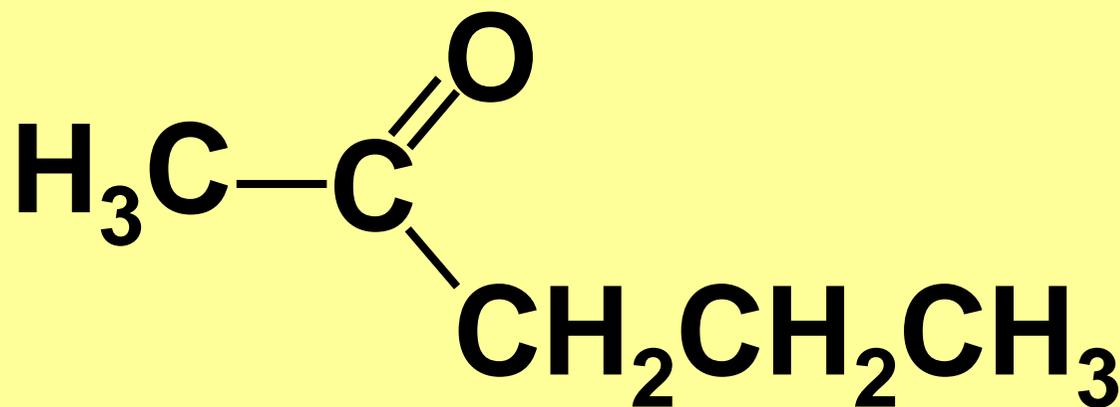
Metanoilbenzene
Benzaldeide o aldeide benzoica



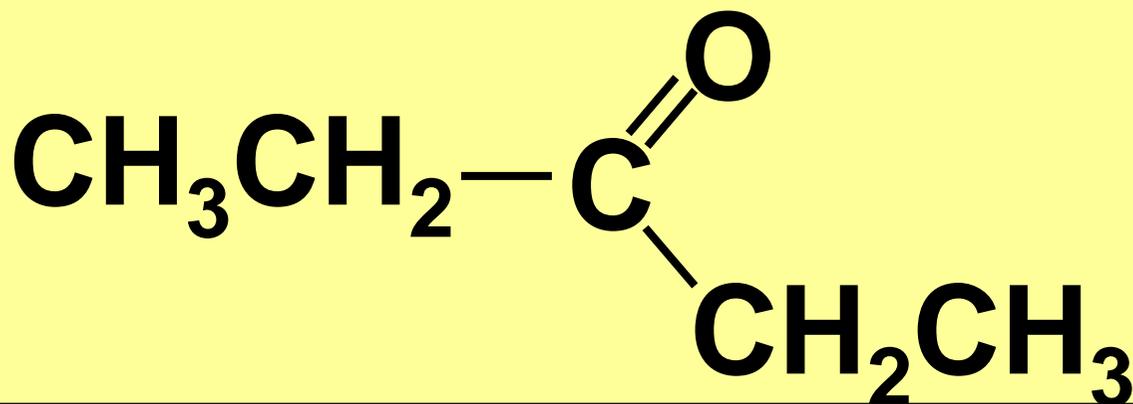
propanone



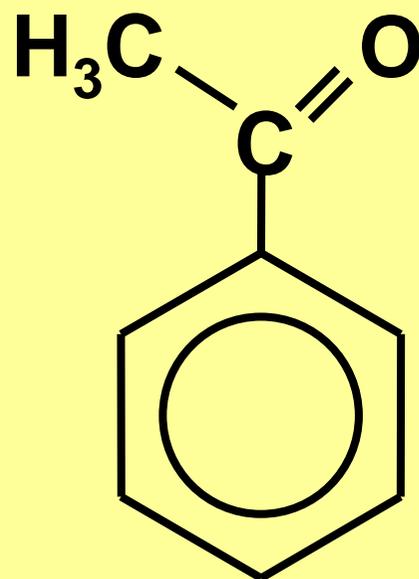
2 - butanone



2-pentanone

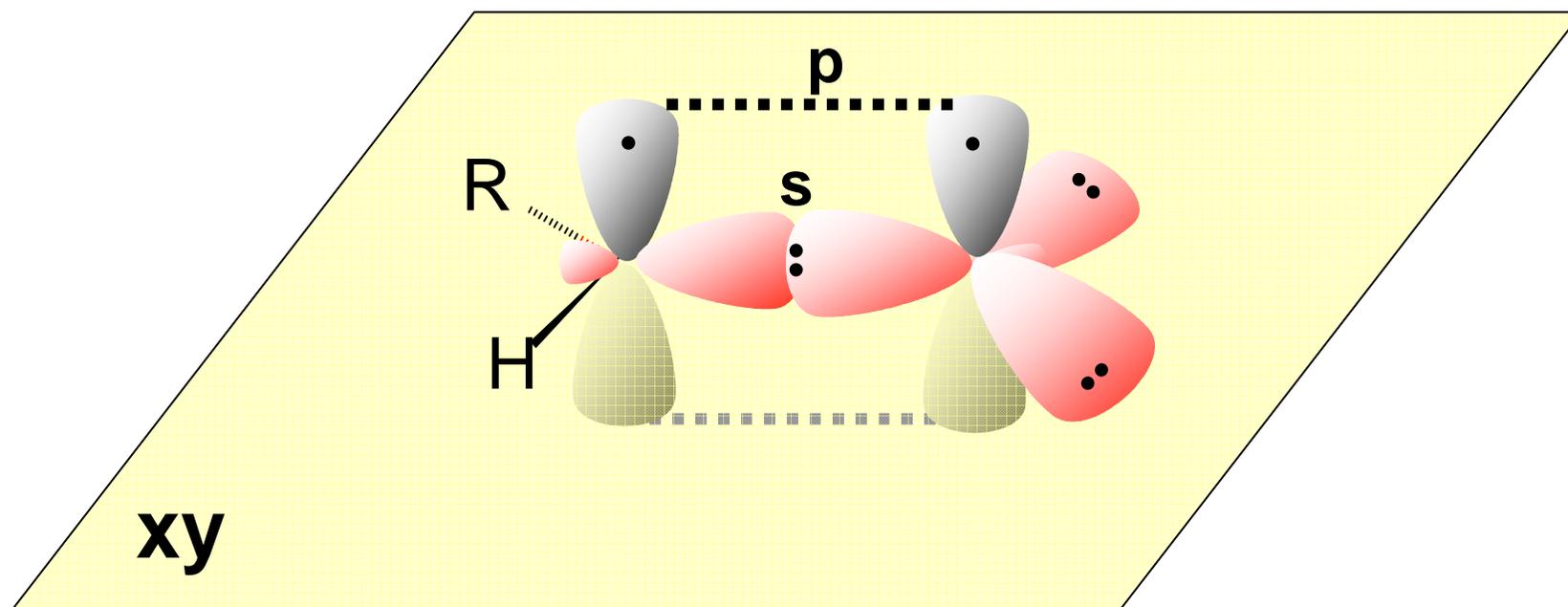
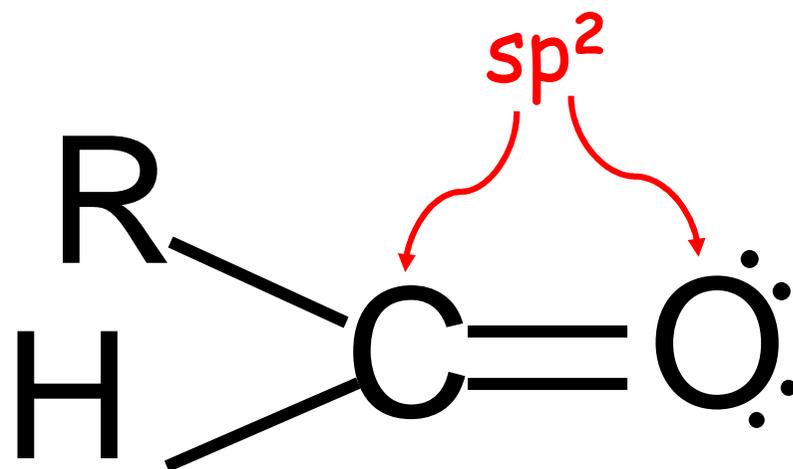


3-pentanone

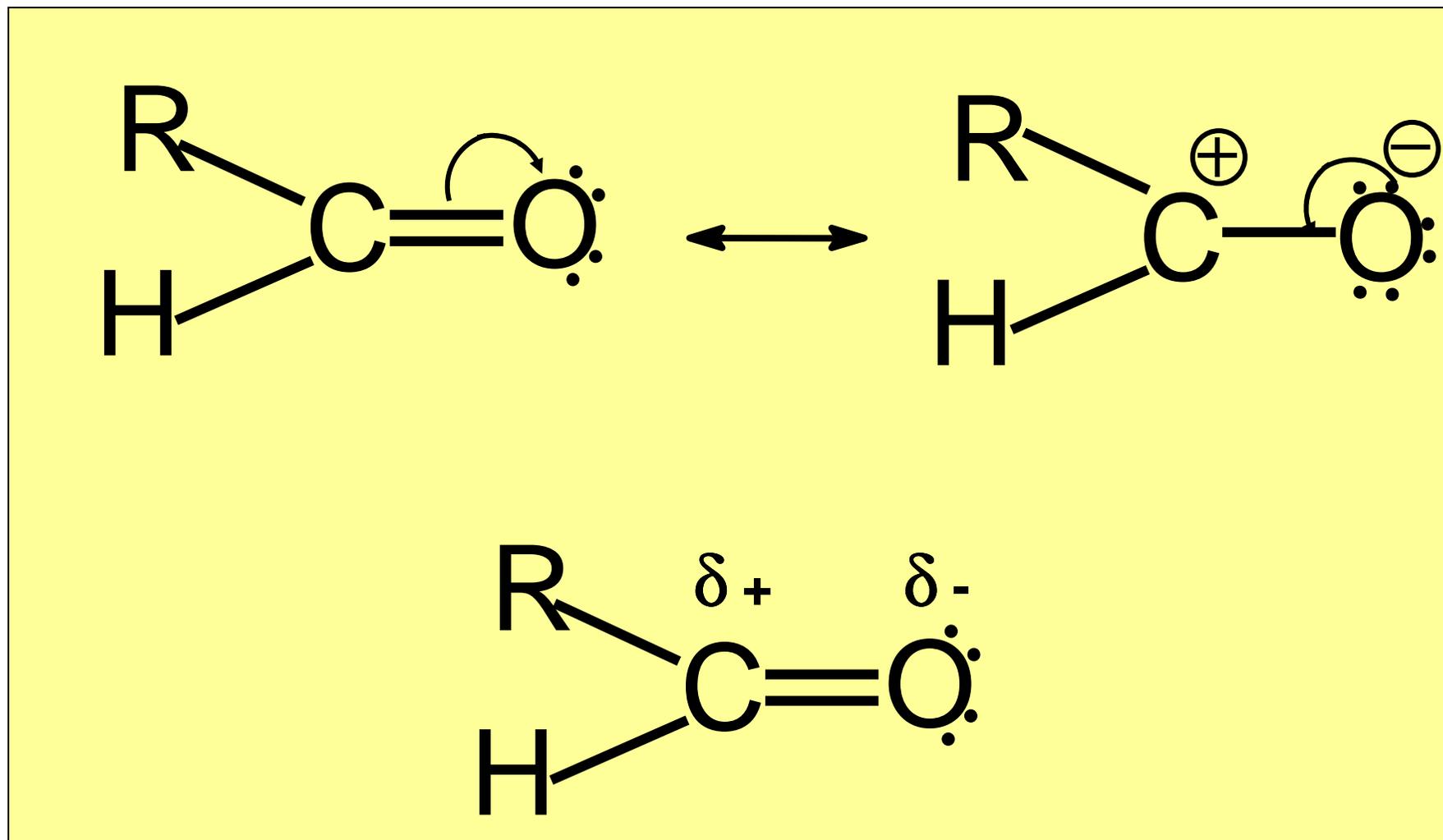


etanoilbenzene

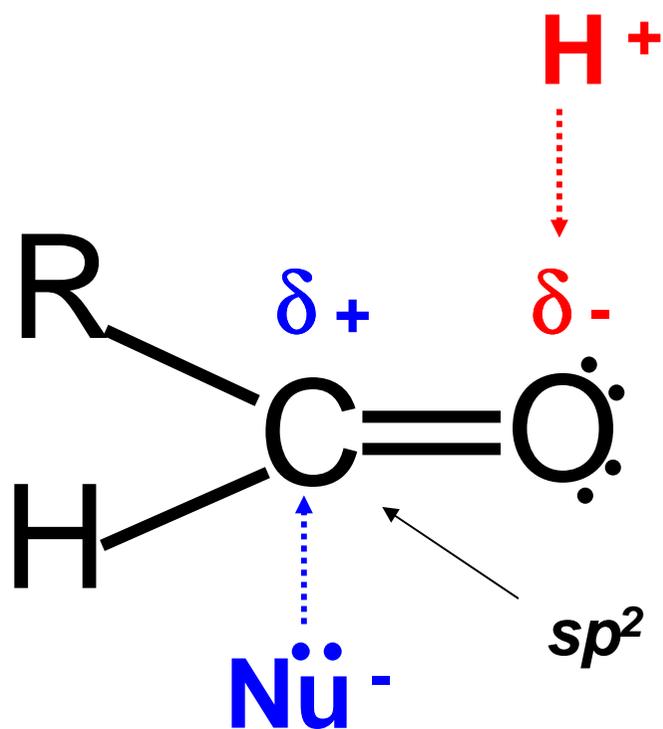
(Derivato acilico del rispettivo idrocarburo)



Forme limite di risonanza

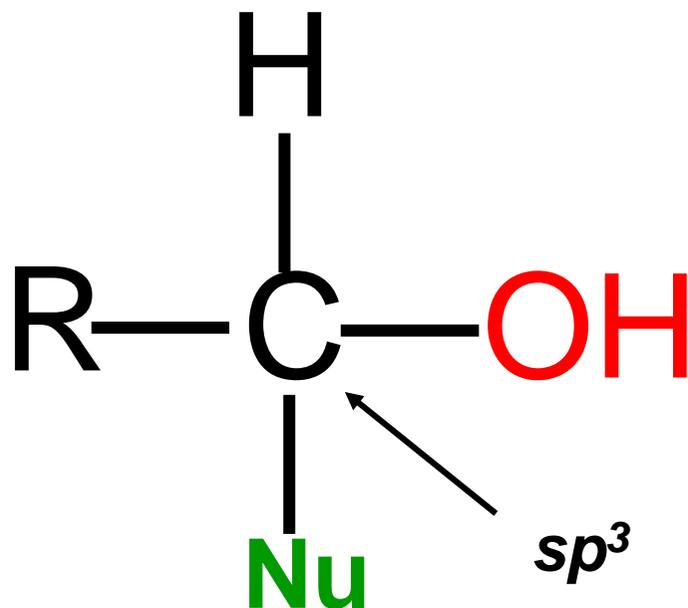


può subire l'attacco di un elettrofilo

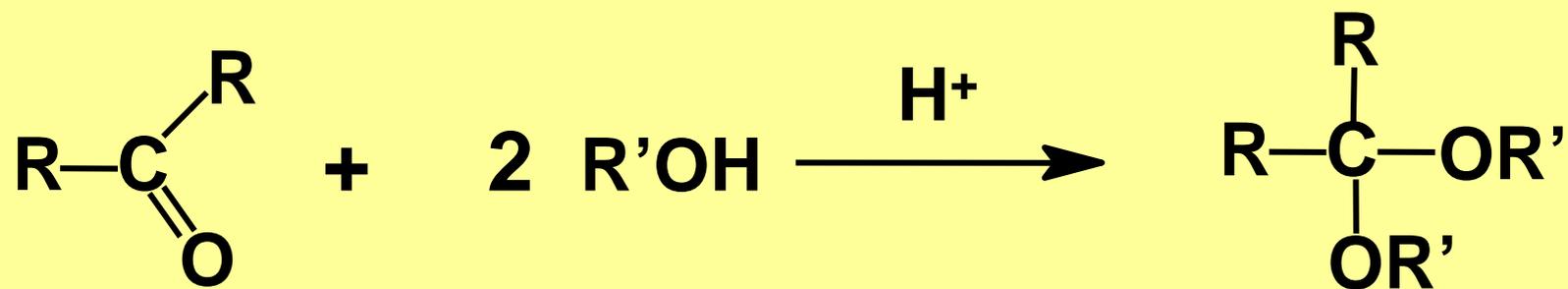
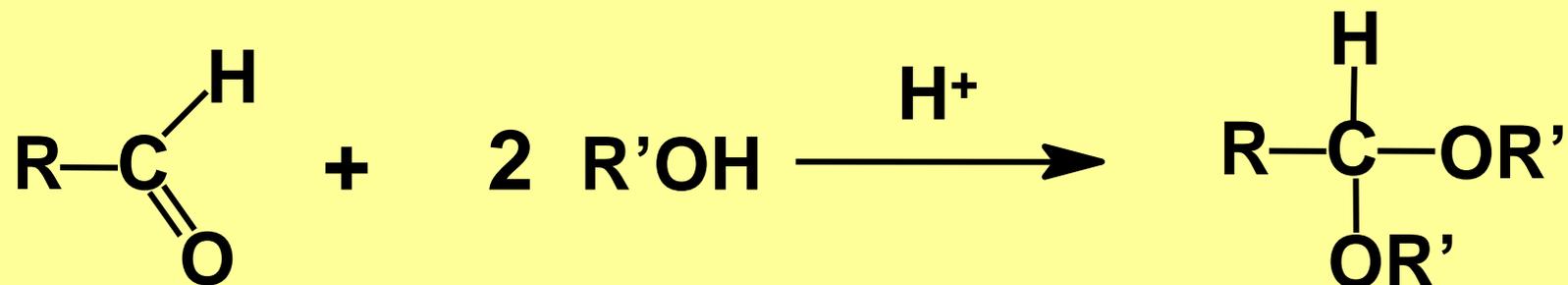


può subire l'attacco di un nucleofilo

Si ottiene un prodotto di addizione nucleofila

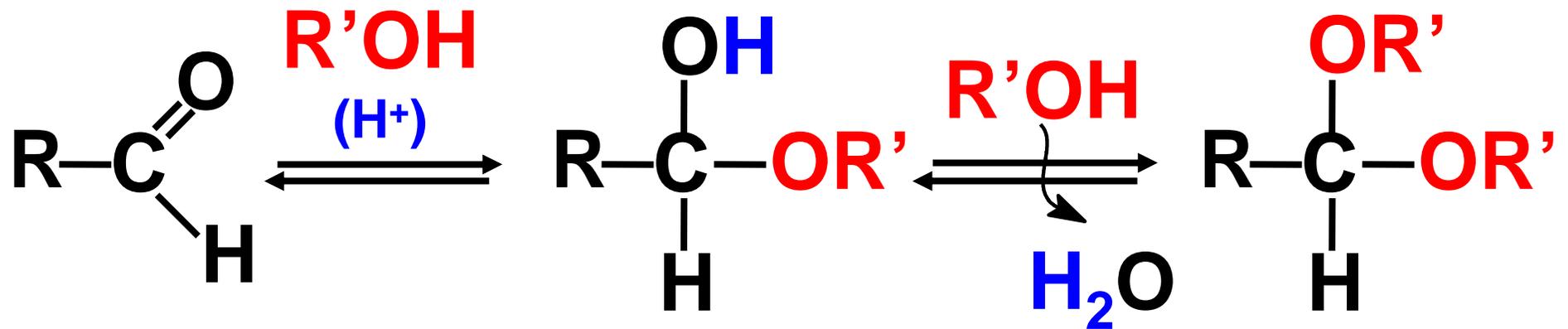


Aldeidi e chetoni danno reazioni di addizione nucleofila



Formazione di acetali

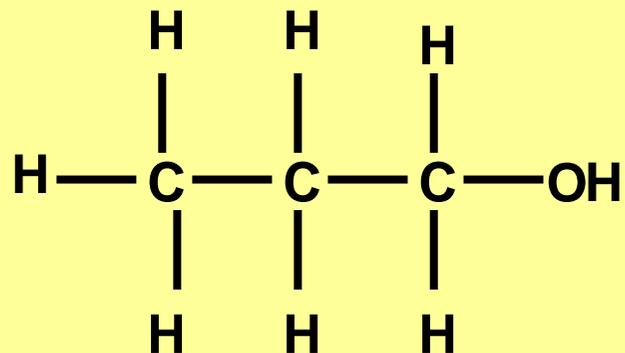
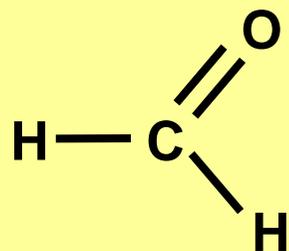
In ambiente acido



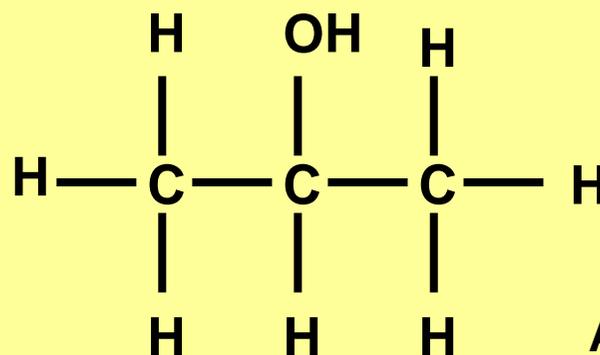
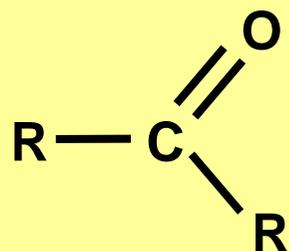
emiacetale

acetale

Riduzione

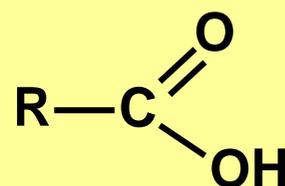
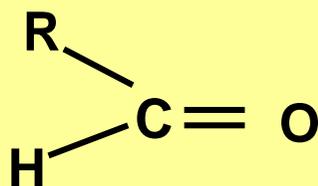


Alcool primario



Alcool secondario

Ossidazione



Acido carbossilico